

# A Review on Implementing Hydrogen Engine- A Case Study of India

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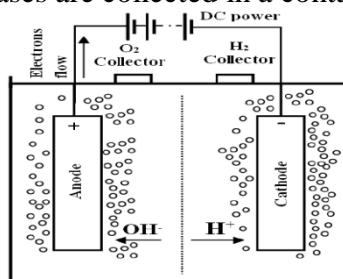
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**ABSTRACT:-** In the present scenery gasoline vehicle are producing measure pollutants like carbon dioxide, oxides of nitrogen, oxides sulphur & particulate matter which are increases global warming. Hydrogen is ultimate fluid overcome automobile pollution by emitting only water. But the challenge in Indian market is production of hydrogen, storages, filling stations and hydrogen engines. Promoting companies to setup their plants with subsidies .Automobile sectors are advised to production of hydrogen engine. A manual engineering technology for setup or replacing filling station in each part of country .Reduction of global warming is reduced 2-5% by increasing use of hydrogen vehicle.

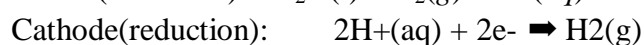
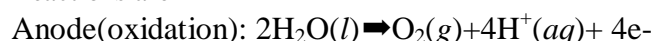
**1. INTRODUCTION:-** Just like note ban it is impossible to stop all the gasoline vehicle and replacing hydrogen vehicle .It is slow and steady process just country like India which development is still going. Hydrogen will be the most widely used fuel in coming 10 to 15 years .Presently cheapest hydrogen is steam reforming of methane where CO<sub>2</sub> reduction is completely not possible .Other procedure like electrolysis, photolytic & thermal process. But still the research is going on producing pure hydrogen. Not only the producing storage handling is little bit use but also cannot detect easily as it is a colorless & odorless gas. Just similar to ICEs of gasoline hydrogen engine are similar way of combustion. Hydrogen engines are higher performances than only at lighter engine load. By increasing load stoichiometric ratio in case of heavy load engine production NO<sub>x</sub> increases which is under the research(Castro, Toledo, & Amador, 2019).(Kalamaras & Efstathiou, 2013)

## 2. PRODUCTION OF HYDROGEN: -

**2.1. ELECTROLYSIS:-** This is very first procedure for the production of hydrogen by decomposition of water into oxygen and hydrogen where current supply starts from 1.23v.process having one cathode and anode both are supplied with electric current, when pure water are supplied to the chamber +<sub>ve</sub> ion of H<sub>2</sub> will passes towards cathode and upper part of the chamber are fitted with a container to store hydrogen. Another part is -ve ion of O<sub>2</sub> is attracted towards anode O<sub>2</sub> gases are collected in a container.



Reactions are



But still the research is going on for better electrode, getting pure water, voltage stability for better accuracy. It also helps in the production of oxygen also consideration about better oxygen.(Carmo & Fritz, 2013)

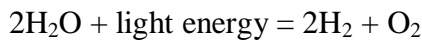
2.2. PHOTOLYTIC PROCESS

2.2.1 **Photoelectrochemical Hydrogen Production (Using Solar Power to Directly Split Water):** In this process photoelectrochemical material (semiconductor) dissociates water into hydrogen and oxygen by using sunlight. PEC materials are same those are used in photovoltaic solar cell. PEC directly convert solar energy into chemical energy. This is good technology for production hydrogen by using solar energy instead of electricity but the rate of production hydrogen is very low. It is only applicable in day time. (Alfaifi, Ullah, Alfaifi, & Tahir, 2018)

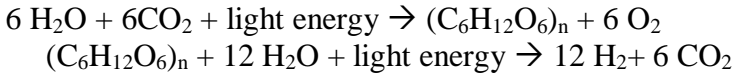


2.2.2 **Biological Hydrogen Production (Photobiological Water Splitting):** These are several types are described bellow.

2.2.3 **Direct biophotolysis:** In this process fresh water algae are used for H<sub>2</sub> and O<sub>2</sub> production by using solar energy directly converted into chemical energy. Also seawater algae are used for the H<sub>2</sub> production. *Chlamydomonas reinhardtii* is the microbe commonly used microbe, *Scenedesmus obliquus* and *Chlorella fusca* are used. (Hallenbeck et al., 2019)

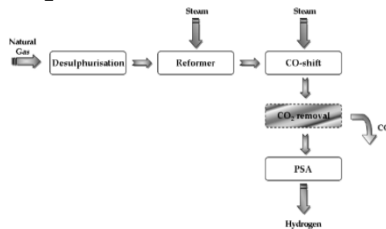


2.2.4 **Indirect biophotolysis:** In this process H<sub>2</sub> and O<sub>2</sub> are produced two steps starting form photosynthesis to sugar formentation. Cyanobacteria from the species of *Anabaena*, *Oscillatoria*, *Calothrix* and *Gloeocapsa* are mainly used. (Alfaifi et al., 2018)

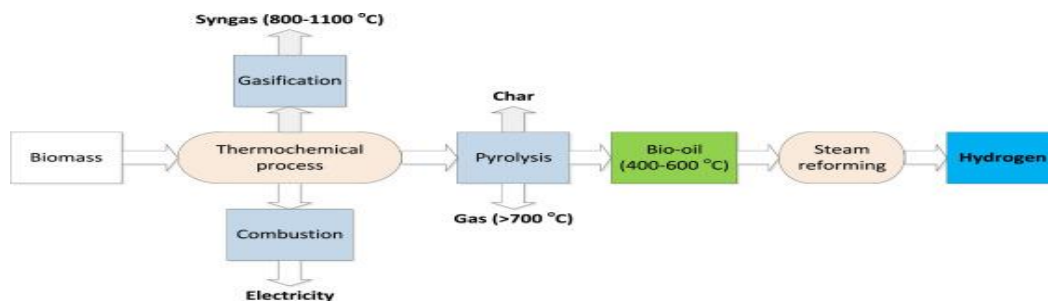


2.3. THERMAL PROCESS

2.3.1. **Distributed Natural Gas Reforming:** This is most widely used hydrogen production. A large scale of production of hydrogen in USA in this process. 700°C-1000°C steam is passed to methane for producing hydrogen under a pressure of 3-25 bar. (Kalamaras & Efstathiou, 2013)



2.3.2. **Bio-Derived Liquids Reforming:** these are the procedure for hydrogen production on sites, in this biomass by process of thermo chemical produced into bio-oil then by steam reforming to convert into hydrogen (“Bio-Derived Liquids to Hydrogen Distributed Reforming Working Group ( BILIWG ), Hydrogen Separation and Purification Working Group ( PURIWG ) & Hydrogen Production Technical Team Research Review 2007 Annual and Merit Review Reports,” 2007)



STORAGE : The six basic hydrogen storage and methods. The gravimetric density  $p_m$ , volumetric density  $p_v$ , working temperature T and pressure P are listed. RT = room temperature at 25°C .(Abe, Popoola, Ajenifuja, & Popoola, 2019)

Storage method	$p_m$ [mass%]	$p_v$ [kg H <sub>2</sub> m <sup>-3</sup> ]	T [°C]	P [bar]	Phenomena and remarks
High pressure gas cylinder	13	<40	RT	800	Compressed gas ( molecule of H <sub>2</sub> ) in light weight composite cylinders(tensile strenght=2000MPa)
Liquid hydrogen in cryogenic tanks	Site dependent	70.8	-252	1	Liquid H <sub>2</sub> , contious loss of few % of hydrogen per day at RT
Adsorbed hydrogen	=2	20	-80	100	Physisorption ( molecule of H <sub>2</sub> ) on materials , e.g. carbon with very large specific surface area, fully reversible
Absorbed on interstitial in a host material	=2	150	RT	1	Hydrogen (atomic H) intercalation in host metals, metallic hydrides working at RT are fully reversible
Complex compound	<18	150	100	1	Complex compound ([ALH <sub>4</sub> ] <sup>-</sup> or [BH <sub>4</sub> ] <sup>-</sup> ), desorption at elevated temperature , adsorption at high pressure.
Material and complexes together with water	<40	>150	RT	1	Chemical oxidation of metal with water & liberation of H <sub>2</sub> , not directly reversible .

3. TANKS : First hydrogen tank was 700bars (70MPa,10000 Psi) was introduced in 2014.further recently Hy-Can tank was constructed for the 1 liter,10 bar format. A classification of hydrogen tank are metal tank (steel/aluminium) where aluminium/glass having 263bar,steel/carbon 299bar.Later composite tank fiberglass/aramid or carbon fiber with metal liner(Al & steel) such as Aluminum/glass-305bar, Aluminum/aramide -438bar, Al/Carbon -700bar again only introduction of carbon fiber with polymer liner of 700bars.(Zhang, Lv, Kang, & Zhou, 2019)

4. **Hydrogen filling station :** In India there should be primarily hydrogen station start from the urban areas by just rearrange the setup the petrol pump.(Thesen & Langhelle, 2009)

How to construct a hydrogen station ?

Layout for each and every part of staion , it must have for both light and heavy vehicle. Basically 2 types of hydrogen are available H70(10000Psi) and H35 (5000Psi) are accodngly specialised engine . Component of a stations are

Hydrogen storage equipment: Based on location and capacity, hydrogen can be stored as a liquid, a low pressure & high pressure gas. Tank must follow ASME standard.

Compressor : Compressor are used to reduce volume and increase pressure (replenish the buffer strength), liquid and low pressure gas hydrogen station may use multiple compressors.

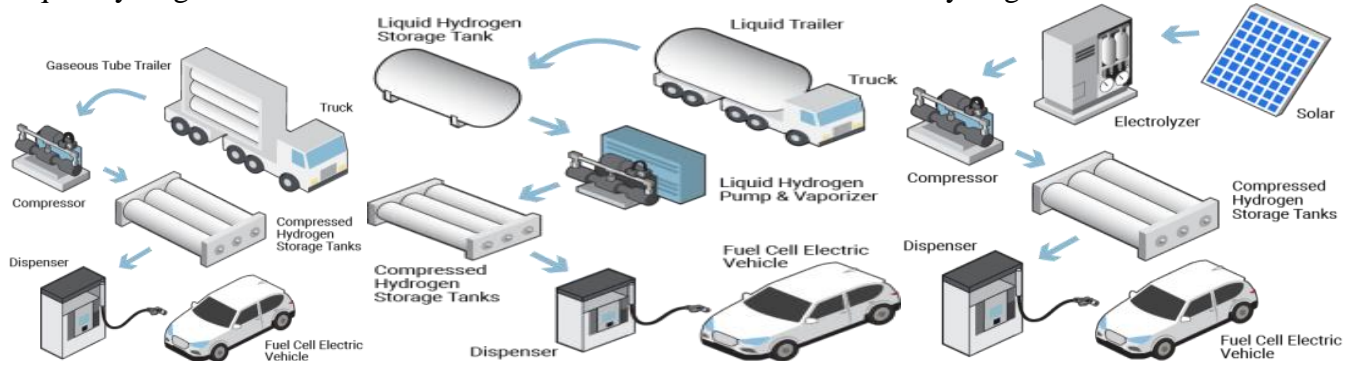
Chiller : Hydrogen needs cooling, not to exceed the threshold temperature of industry standard fueling protocol.

Dispenser : Must have one or more according to hydrogen requirements in the locality.

Liquid hydrogen

Gaseous

Onsite hydrogen



**5. Hydrogen Engine :**

Since 1976 hydrogen engine has had a great impact. Hydrogen engine has 3 primary obstacles: safety, infrastructure and cost. It is lighter than air, weight reduction as compared to gasoline engine. If any hole in the cylinder, it does not explode simply escapes and dissolves in the atmosphere. Hydrogen is pretty much higher than gasoline engine as high stress materials are used. (Castro et al., 2019) (Deheri, Acharya, Thatoi, & Mohanty, 2020)

**6. Function :**

From the filling station hydrogen is received with a receiver pipe with a special arrangement of intake valve where a spring arrangement of rod opens. After filling the rod gets its original position. Hydrogen is stored in the tank then supplied to the engine by mixing with air in the carburetor or fuel injector to enhance the combustion process. For a hydrogen engine, emission is only water, but in case of high engine load as a rich mixture, also NOx is emitted. (Liu et al., 2019)

Just like a gasoline (petrol) engine, a hydrogen engine also works on 4 working strokes:

**Suction :** intake of air and fuel (in gaseous form) to the engine. The piston moves from TDC to BDC, intake valves are opened, after it reaches in the BDC, suddenly the intake valve closes by regulating with the crankshaft.

**Compression :** In this stroke, the piston moves from BDC to TDC. The A/F mixture gets compressed to catch fire easily, the spark plug will give a spark to the engine to start the combustion process. Hydrogen engines are like petrol engines, they cannot ignite themselves by compression.

**Expansion or working stroke :** The piston again moves from TDC to BDC and the valve remains closed. Combustion power (hydrogen molecules) gives power to the piston move downward.

**Exhaust :** piston moves from BDC to TDC at the same time, the exhaust valve gets opened so the combusted gases are emitted through it to the tail pipe (cylinder), generally a hydrogen engine only emits water.

Emission are  $2H_2 + O_2 \rightarrow 2H_2O$

During high engine  $H_2 + O_2 + N_2 \rightarrow H_2O + NO_x$

**7. Adaption of existing engine :**

Materials for adapting an existing engine are valve seats and hardened valves, connecting rods, non-platinum tipped spark plug, higher voltage ignition coil, fuel injector are designed for gases instead of liquid, larger crankshaft damper, modified intake manifold, stronger head gasket, positive pressure supercharger and high temperature engine oil which is approximately 1.5 times higher in cost. Stoichiometric air fuel ratio was about 34:1. The maximum output is 15% higher than gasoline. As such a datation in the existing engine hydrogen engines are bigger than gasoline engines. (Kakooe, Bakhshan, Motadayen, & Gharehghani, 2018)

## 9 CONCLUSION:

This article gives overall view of hydrogen production & storage, modification of engine which is completely beneficial than gasoline engine and it is a next generation green engine. As in India, it is not a big issue to modify existing engines. To compete in international market for petroleum very useful steps should be hydrogen engine, only initial setup for hydrogen plant is pretty high cost. Still only one big research is reduction of NOX at high engine load. This may be overcome by introducing hydrogen based sensor which provides a consistent A/F mixture to engine in coming years. Hope fully next coming 10 years hydrogen engine will be seen in India.

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