

**Experimental Investigation On Synthesis And Biological Evaluation Of
Pyrazolidine-3,5-Dione****Mir Mohammad Shahroz**Research Scholar in Pharmacy, Faculty of Pharmacy, Sri Satya Sai University Of
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University Of Technology And Medical Sciences, Sehore, MP**ABSTRACT**

There has been a most concerning issue of bacterial opposition as far back as the improvement of anti-bacterial specialists. The present research work centers around the microwave helped dissolvable less synthesis joined with ordinary stirring and refluxation techniques to frame some novel subbed 4-quinolone pyrazolidinedione derivatives. The characterization of n=9 derivatives was done utilizing I.R, ¹H NMR and mass ghastly analysis. The rate yield of definite mixes was seen as 22.15 to 69.68. Purity of the mixes was checked by utilizing TLC and natural analysis. These mixes demonstrated a significant anti-bacterial movement against *S. aureus*, *B. subtilis*, *Klebsiellapneumoniae* and *Proteus vulgaris* and anti-inflammatory action utilizing Invitro testing techniques contrasted with Ciprofloxacin, Amoxicillin and Ibuprofen respectively.

Keywords: Quinolone, pyrazolidine-3,5-dione, antibacterial agents, anti-inflammatory agents, antimicrobial

INTRODUCTION

Pyrazolone is a five - membered lactam ring compound containing two nitrogen iotas and ketone in a similar atom. Lactam structure is a functioning nucleus in pharmacological action. Pyrazolone is a functioning moiety as a pharmaceutical fixing, particularly in the class of nonsteroidal anti-inflammatory agents utilized in the treatment of joint inflammation and other musculoskeletal and joint issue.

The quick ascent in microbial resistance to the customary antibiotics has required a proceeding with look for new classes of compounds with novel methods of antimicrobial movement.

Albeit an enormous number of anti-bacterial agents are accessible in the market for the treatment of fatal bacterial infections but since of the developing resistance issue to these agents, the requirement for new antibacterial agents is likewise developing proportionately. One of the most utilized anti-bacterial classes of drugs is fluoroquinolones.

Inflammation is a reaction of the tissue to an infection, disturbance or outside substance and is a piece of the host barrier system. The inflammatory procedure includes a progression of occasions that can be inspired by various improvements (for example infectious agents, ischemia, antigen-antibody association and thermal or other physical wounds).

The 4-quinolone antimicrobials have various focal points over different classes of antimicrobial agents. They have wide range of action, all around retained orally, have generally long serum half lives and insignificant toxicity. The as of now utilized quinolone derivatives are known to have a few drug associations and unfriendly reactions.

The title compounds were prepared through several steps as depicted in Fig 1.

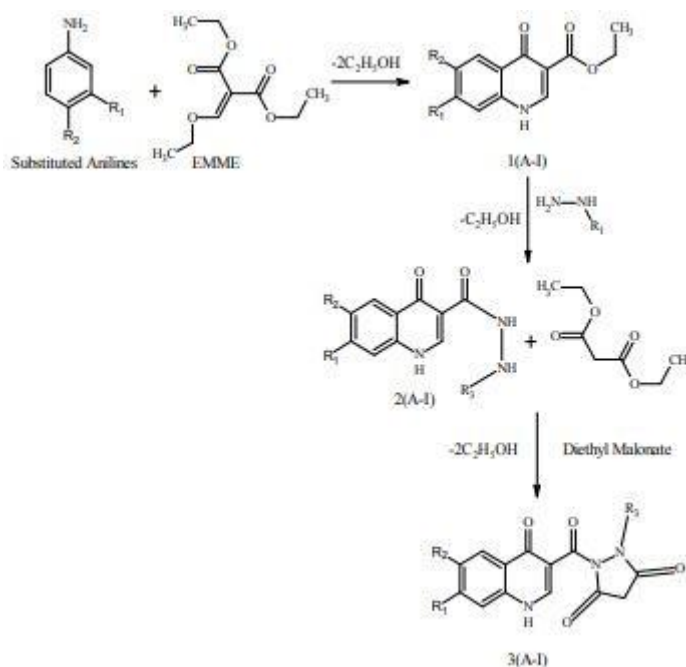


Figure 1: Synthesis of the title compounds

LITERATURE REVIEW

MANISH GUPTA, NEERAJ UPMANYU, SOMA PRAMANIK, CHANDRA KISHORE TYAGI AND AMOL CHANDEKAR (2011) A progression of novel 1-(2-methyl-4-oxo-1,4-dihydro quinoline-6-carbonyl)- 2-(subbed phenyl)- pyrazolidine-3, 5-diones [5A-5H] were incorporated with various sweet-smelling hydrazides and assessed for antibacterial and antifungal movement. The structures of the compounds were affirmed by nitrogen analysis, FT-IR, 1H NMR and mass ghostly data. The antimicrobial movement of combined compounds were analyzed against two gram positive bacteria (*S. aureus*, *B. subtilis*), two gram negative bacteria (*E. coli*, *S. species*) and two organisms (*C. albicans* and *A. niger*) utilizing soup dilution

technique. A portion of the blended compounds displayed gentle to direct antibacterial and antifungal action.

SuvarnaKini and A. M. Gandhi(2008) The 1,3,5-trisubstituted-2-pyrazolines were integrated by refluxing isoniazid with different subbed diarylchalcones in N,N-dimethylformamide at 120-140°. The physical and phantom data, for example, M.P., Rf, natural analysis, IR, NMR and Mass was gotten for the integrated compounds and the structures were affirmed. The screening of the orchestrated compounds for antimicrobial movement was performed against *Staphylococcus aureus*, *Bacillus subtilis*, *Pseudomonas aeruginosa*, *Escherichia coli* and *Aspergillusniger*.

Annasaheb B. Jagnar, Shital D Gaikwad, PriyankaM.Wavare (2016) various subbed pyrazolidin-3-one, aryl oxadiazole and mercaptooxadiazoles are known for their biological significance like anti-bacterial, antitubercular, antioxidant and antiinflammatory action. The title compounds has been combined from diclofenac acid and ofloxacin responds with ethanol in nearness of acid to give hydrazide which on further response with Ethylacetoacetate gives subbed pyrazolidin-3-one which experiences Mannich response gives Pyrazolidin-3-one. Acid hydrazide on treatment with Aromatic acids in nearness of Phosphorus oxychloride gives aryl oxadiazoles. Mercaptooxadiazoles orchestrated utilizing acid hydrazide and carbon disulphide. The recently blended compounds have been described by IR, ¹H NMR and CHN analysis.

Mohamed A. H. Ismaila, Dalal A. Abou El Ella, Khaled A. M. Abouzid, and MaiyJaballah (2012) aganists as putative drugs for the treatment of γ . In the mission for novel PPAR type 2 diabetes, another arrangement of 2-pyrazolin-5-one and pyrazolidine-3, 5-dione derivatives, were planned and integrated, as analogs to the antidiabetic thiazolidinedione agents (TZDs). Broad sub-atomic displaying agonist's theory andystudies for the planned particles were performed; including their comparefit thinks about on the produced and approved PPAR γ their sub-atomic docking on the coupling destinations of the 3D structure of the PPAR receptors.

Deepak Swarnkar, Rakshit Ameta, and Ritu Vyas (2014) A progression of subbed 1,3,4-oxadiazole derivatives (3a–f) and (6a–f) have been integrated from diphenylacetic acid hydrazide under microwave illumination in different response conditions. The structures of the integrated compounds were doled out based on basic analysis, IR, and ¹H NMR. These focused on compounds have been tried for their antibacterial and antifungal exercises contrasted with ampicillin and griseofulvin as standard drug. Compounds 3a, 3e, 3f, 6c, 6d, 6e, and 6d exhibited the greatest antibacterial exercises while 3b, 3c, 3d, 3e, 6a, 6d, and 6e exhibited the most extreme antifungal exercises

MATERIAL AND METHODS

The chemicals and reagents utilized in this were of AR and LR grade. They were secured from SpectroChem, Hi-Media, Merck, Sigma Aldrich and Ranbaxy.

Synthesis of substituted ethyl esters of 4-oxo-1,4- dihydroquinoline-3-carboxylic acids [1A-1I]

Equimolar quantities of subbed anilines (0.01 mol) and diethyl (ethoxymethylene) malonate (EMME) (0.01mol), (practically boring fluid, b.p. 279-281 °C) were taken in a measuring glass under solvent-free condition. At the point when an unmistakable arrangement was acquired by shaking, it was lighted under microwave for min at high force (900 watts) while shaking the blend at interims of 30 sec. The subsequent arrangement was inevitably changed over to pale mass after cooling with ice. This was trailed by washing it with CH₃CO and the buildup was recrystallised utilizing N, N-Dimethyl formamide (DMF) as solvent. Dissolving focuses, rate yields and time of micro wave light of all 1A-1I compounds are given in table 1(a).

Table 1: Substituted ethyl esters of 4-oxo-1,4-dihydroquinoline-3-carboxylic acids

Parameters	Codes								
	1A	1B	1C	1D	1E	1F	1G	1H	1I
R ₁	Chloro	H	H	H	H	H	H	H	Chloro
R ₂	Fluoro	H	F	Cl	Br	I	CH ₃	NO ₂	H
R ₃	H	H	H	H	H	H	H	H	Ph
% Yield	50.39	46.96	52.07	40.15	40.05	47.23	45.86	46.30	50.39
Melting Point (°C)	290-295	270-275	296-298	310-315	318-320	306-308	295-301	260-261	290-292
Time of Irradiation (min)	1-1.5	1.5-2	4.5-5	2-3	2-3	2-3	1-1.3	0.4-0.5	1-1.5

Synthesis of substituted 4-oxo-1,4- dihydroquinoline-3-carboxylic ethyl ester hydrazide [2A-2I]

Equimolar quantities of ethyl esters of 4-oxo-1,4-dihydroquinoline-3-carboxylic acids and hydrazine hydrate was taken in nearness of DMF as a solvent in an iodine jar. The response blend was kept for stirring utilizing attractive stirrer for 16 to 18 hours. The reasonable arrangement acquired was poured in ice cold water with steady stirring. The arrangement was kept at room temperature for 20 to 24 hours to get accelerates of hydrazide. The strong hastens were separated, dried and recrystallised utilizing DMF-ethanol (1:1) blend. Rate yield and liquefying purposes of all the (2A-2I) derivatives are given in table 2.

Table 2: Substituted 4-oxo-1,4-dihydroquinoline-3-carboxylic ethyl ester hydrazides

Parameters	Codes								
	2A	2B	2C	2D	2E	2F	2G	2H	2I
R ₁	Chloro	H	H	H	H	H	H	H	Chloro
R ₂	Fluoro	H	F	Cl	Br	I	CH ₃	NO ₂	H
R ₃	H	H	H	H	H	H	H	H	Ph
% Yield	56.96	48.25	54.47	47.97	47.51	58.39	55.73	48.43	13.59
Melting Point (°C)	>350	>350	>350	>350	>350	>350	>350	>350	>350

Synthesis of 7-chloro-6-fluoro-4-oxo-N'-phenyl-1,4-dihydroquinoline-3-carbohydrazide [2I]

Equimolar quantities of ethyl 7-chloro-6-fluoro-4-oxo-1,4-dihydroquinoline-3-carboxylate and phenyl hydrazine was taken in a 50ml of container. Presently the suspension got is exposed to microwave irradiation at high force (900 watts) for 7-8min with stirring at an interim of 30 sec. The red shaded arrangement was cooled in ice. The blend was washed with ethanol to give accelerates in the wake of keeping aside the reasonable answer for 17 hours. The strong encourage got was separated and recrystallised utilizing ethanol. Rate yield and softening point for the derivative is given it table 2.

Synthesis of substituted 4-quinolone containing pyrazolidine-3,5-dione derivatives [3A-3I]

An equimolar quantities of subbed 4-oxo-1,4-dihydroquinoline-3-carboxylic ethyl ester hydrazide (2A-2I) and diethyl malonate was taken in DMF ethanol (1:1) arrangement and acidic acid (1ml) was added to it in a round bottom flask. The blend was refluxed for 6-7 hrs. The response blend was left in open dish for 2-3 hrs. The solid precipitate framed was sifted, dried and recrystallized utilizing ethanol. Rate yield and melting points for all the blended derivatives are exhibited in table 3.

Table 3: Substituted 4-quinolone containing pyrazolidine-3,5-dione derivatives

Parameters	Codes								
	3A	3B	3C	3D	3E	3F	3G	3H	3I
R ₁	Chloro	H	H	H	H	H	H	H	Chloro
R ₂	Fluoro	H	F	Cl	Br	I	CH ₃	NO ₂	H
R ₃	H	H	H	H	H	H	H	H	Ph
% Yield	22.11	48.48	35.5	40.41	69.68	49.63	50.12	60.75	45.09
Melting Point (°C)	>350	>350	>350	>350	>350	>350	>350	>350	>350

Biological Activity

- 1) Anti-bacterial movement: Agar plates were set up by emptying supplement agar media into the petridishes and every one of them were immunized with a specific microorganism, similar to gram-positive (*S. aureus* and *B. subtilis*)

and gram-negative (*Klebsiella pneumoniae* and *Proteus vulgaris*). After agar was solidified, cups were made in the supplement agar. The Anti-microbial test drugs 3A-3I is put in the cups (1000µg). The drug diffuses through the agar around the cup. The plates are brooded at a temp of 37°C for 24hrs for bacterial culture. The standard reference drugs utilized in the antibacterial screening were Ciprofloxacin and Amoxicillin (100 µg/ml).

- 2) Anti-inflammatory action (in vitro Inhibition of albumin denaturation): The test compounds were broken up in least measure of dimethyl formamide (DMF) via completing sonication for 10-15mins and diluted with phosphate buffer (0.2M, pH 7.4). The last concentration of DMF in all arrangements was under 2.5%. Test arrangement (1ml) containing diverse concentration of integrated compounds [3A-3I] was blended in with 1ml of 1mg/ml albumin arrangement in phosphate buffer and hatched at 27°±1°C for 15 min. Denaturation was prompted by keeping the response blend at 60°±1°C in water shower for 10-20 min. in the wake of cooling, the turbidity was estimated at 660nm in U.V spectrophotometer. The rate inhibition of denaturation was determined from control where no incorporated compounds were included and looked at against standard (Ibuprofen).

RESULT

Synthesis of 1A-1I was done utilizing microwave irradiation under solvent free condition, physical data. The yields were seen as standard with that of the ordinary strategies which utilizes natural solvents. Along these lines this technique is an efficient and cost sparing strategy. The technique was additionally seen as valuable for synthesis of phenyl hydrazide under solvent less condition. Essential analysis of all the last 3A-3I compounds demonstrated that they were exceptionally unadulterated; purity was additionally checked utilizing TLC. An extensive antibacterial action was appeared by all the blended derivatives contrasted with standard drugs.

Compound 3D was seen as increasingly compelling against *S. aureus* while 3B and 3G have indicated great movement against *B. subtilis*. 3G has given some movement against *Klebsiella pneumoniae*. Compound 3C has demonstrated great movement against *Proteus vulgaris*. From the invitro anti-inflammatory action contemplates it was uncovered that 3B and 3D gave similarly better inhibition to albumin denaturation with increment in concentration from 0.2mg/ml; to 1.0 mg/ml.

CONCLUSION

It very well may be inferred that blended combined heterocyclic ring show great antifungal activity as well as give great antibacterial activity too. The viability of the integrated compounds as antibacterial was seen as more prominent, when contrasted

with antifungal activity. Compound with nitro substitution on benzene ring is progressively best. These compounds can be considered as lead molecules for future examinations.

REFERENCES

- [1] Manish Gupta, NeerajUpmanyu, Soma Pramanik, Chandra Kishore Tyagi and AmolChandekar(2011), 'SYNTHESIS AND ANTIMICROBIAL EVALUATION OF 3,5- PYRAZOLIDINE-DIONE SUBSTITUTED 4-QUINOLONE DERIVATIVES', International Journal of Drug Development & Research
- [2] Kini S. and Gandhi A. M., Novel 2-pyrazoline derivative as potential anti-bacterial and antifungal agents, Indian J. Pharm Sci., 2008, 70, 105-108.
- [3] Annasaheb B. Jagnar, Shital D Gaikwad, PriyankaM.Wavare (2016) 'Design, Synthesis and Evaluation of Some Novel Pyrazolidine-3- One, Aryl Oxadiazole and MercaptoOxadiazole Derivatives of Biological Interest', Human Journals, February 2016 Vol.:5, Issue:3
- [4] Mohamed A. H. Ismaila, Dalal A. Abou El Ella, Khaled A. M. Abouzid, and MaiyJaballah (2012), 'DESIGN, SYNTHESIS AND VIRTUAL SCREENING OF CERTAIN 2-PYRAZOLIN-5-ONE AND PYRAZOLIDINE-3, 5- AGONISTS γ DIONE DERIVATIVES AS POTENTIAL PPAR γ AGONISTS', Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Ain Shams University, El Khalifa ElMaamoon St., 11566, Abbasseya, Cairo, Egypt, 2012; Vol. 3(10): 3746-3757
- [5] Deepak Swarnkar, RakshitAmeta, and Ritu Vyas (2014), 'Microwave-Assisted Synthesis of Some 1,3,4-Oxadiazole Derivatives and Evaluation of Their Antibacterial and Antifungal Activity', OrganicChemistry International Volume 2014, Article ID 694060, 6 pages
- [6] W. S. Han, J. K. Lee, J.-S.Lee, H.-G. Hahn, and C. N. Yoon, "Study of thiazoline derivatives for the design of optimal fungicidal compounds using multiple linear regression (MLR)," Bulletin of the Korean Chemical Society, vol. 33, no. 5, pp. 1703–1706, 2012. View at Publisher · View at Google Scholar · View at Scopus
- [7] G. La Regina, R. Silvestri, V. Gatti et al., "Synthesis, structure-activity relationships and molecular modeling studies of new indole inhibitors of monoamine oxidases A and B," Bioorganic and Medicinal Chemistry, vol. 16, no. 22, pp. 9729–9740, 2008. View at Publisher · View at Google Scholar · View at Scopus
- [8] A. Ramazani and A. Souldozi, "Iminophosphorane-mediated one-pot synthesis of 1,3,4-oxadiazole derivatives," Arkivoc, vol. 2008, no. 16, pp. 235–242, 2008. View at Google Scholar · View at Scopus

[9]A. Husain and M. Ajmal, "Synthesis of novel 1,3,4-oxadiazole derivatives and their biological properties," *Acta Pharmaceutica*, vol. 59, no. 2, pp. 223–233, 2009. View at Publisher · View at Google Scholar · View at Scopus

[10] A. Padmaja, C. Rajasekhar, A. Muralikrishna, and V. Padmavathi, "Synthesis and antioxidant activity of disubstituted 1,3,4-oxadiazole, 1,3,4-thiadiazoles and 1,2,4-triazoles," *Journal of Chemical and Pharmaceutical Research*, vol. 4, no. 1, pp. 294–302, 2012. View at Google Scholar